

Photocycloaddition of Six-Membered Cyclic Enones to Propen-2-yl Isocyanate

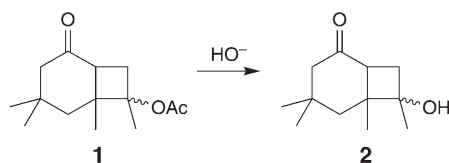
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On irradiation in the presence of propen-2-yl isocyanate (**4**), six-membered cyclic enones **3** are converted into regio- and stereoisomeric mixtures of [2+2] cycloadducts **5–10**; the preferentially formed *HT* products, **5–8**, can be converted into the corresponding bicyclic amines by acid hydrolysis, whereas, under these conditions, the regioisomeric *HH*-isocyanato derivatives undergo a *retro-Mannich* reaction.

1. Introduction. – Photochemistry is a highly valuable tool for modern organic synthesis, as illustrated by one of the most important reaction types, *i.e.*, the stepwise [2+2] photocycloaddition of a cyclic unsaturated ketone to an alkene to give an (annulated) cyclobutane [1]. In this context, it is noteworthy that the synthesis of the diastereoisomeric mixture of 2-oxobicyclo[4.2.0]oct-7-yl acetates **1** by irradiation of isophorone in the presence of propenyl acetate represents one of the, unfortunately, very few procedures in *Organic Syntheses* found under the heading ‘*Photochemical Reaction*’ [2]. Alkaline hydrolysis of these cycloadducts affords a mixture of stereoisomeric alcohols **2** (*Scheme 1*). Interestingly, up to now no such approach has been reported for the synthesis of 7-amino derivatives of these bicycles. One reason for this is that, on irradiation of cyclohexenones in the presence of enamines, electron transfer to the excited enone forestalls any cycloaddition [3]. Moreover, enamides analogous to the enol acetates mentioned before, *e.g.*, *N*-(prop-2-enyl)acetamide, are rather unstable in solution and only accessible by pyrolysis [4], albeit (intramolecular) photoisomerizations of vinylogous amides to azabicyclooctanones have been reported [5]. A very recent photochemical procedure for the synthesis of cyclobutylamines is based on the light-induced cycloaddition of uracil to ethene, followed by controlled degradation of the heterocyclic ring, *i.e.*, the N-atom stemming from the ‘enone’ [6].

Scheme 1



Here, we report that alkenyl isocyanates represent convenient ‘alkene’ partners for such light-induced annelation reactions.

2. Results. – Irradiation ($\lambda = 350$ nm) of a mixture of dihydropyranone **3a** and excess prop-2-enyl isocyanate (**4**) in benzene for 5 h (total enone conversion as monitored by GC) affords mixtures of cycloadducts **5a–10a** (Scheme 2). The constitution and configuration of the photoproducts can be fully assigned by NMR spectroscopy. Differentiation between regioisomers on the one side (for ‘*HH*’-regioisomers the C=O and NCO groups point to the same, for ‘*HT*’-regioisomers towards opposite directions) and defining the relative configuration (*cis/trans*) of the ring fusion, on the other side, directly from the product mixture becomes straightforward from the – easily detectable – splitting pattern of both bridgehead H-atoms (Table). The *exo/endo* assignment of the Me group on the cyclobutane ring stems from NOESY experiments.

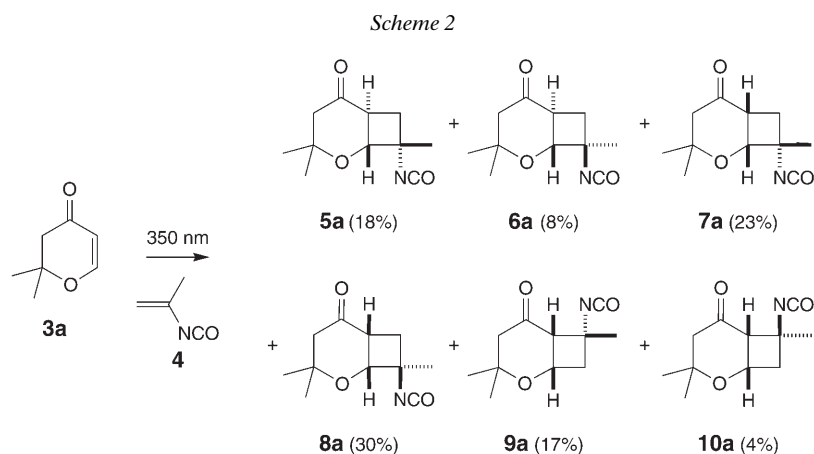
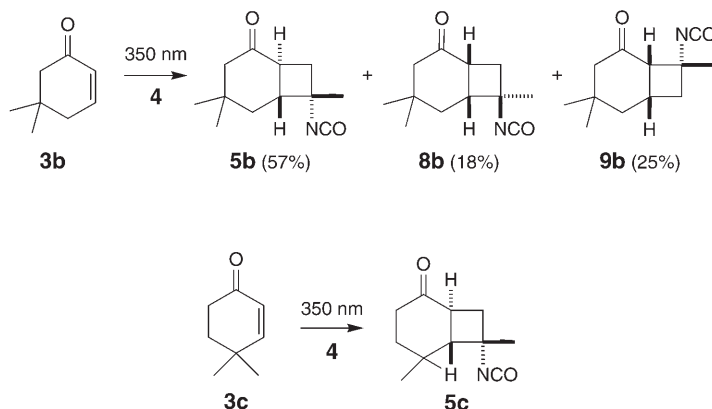


Table. NMR Data (chemical shifts [ppm] and coupling constants [Hz]) of Bridgehead Atoms of Photocycloadducts from **3a** + **4** (in CDCl_3)

	HC–C=O	CH–C=O	HC–O	CH–O
5a	3.24 (<i>ddd</i> , $J = 6.8, 10.8, 11.0$)	51.0	3.62 (<i>d</i> , $J = 11.0$)	80.1
6a	2.65 (<i>ddd</i> , $J = 6.8, 10.8, 11.2$)	48.2	3.82 (<i>d</i> , $J = 11.2$)	82.3
7a	2.78 (<i>ddd</i> , $J = 4.9, 9.0, 9.5$)	37.7	4.23 (<i>dd</i> , $J = 3.3, 4.9$)	76.1
8a	3.08 (<i>ddd</i> , $J = 6.6, 7.0, 9.9$)	38.4	4.43 (<i>dd</i> , $J = 2.0, 6.6$)	77.3
9a	2.86 (<i>d</i> , $J = 5.2$)	52.9	4.48 (<i>ddd</i> , $J = 1.5, 5.2, 6.1$)	64.9
10a	3.08 (<i>d</i> , $J = 5.2$)	54.9	4.55 (<i>ddd</i> , $J = 1.3, 5.2, 6.2$)	69.6

Under similar conditions, irradiation of **3b** affords a mixture of three cycloadducts, **5b**, **8b**, and **9b**, in 57, 18, and 25% relative yield, respectively. In contrast, irradiation of **3c** in the presence of excess **4** gives cycloadduct **5c** selectively, but the rate of conversion to product(s) is *ca.* 2–3 times slower than for the former two enones (Scheme 3).

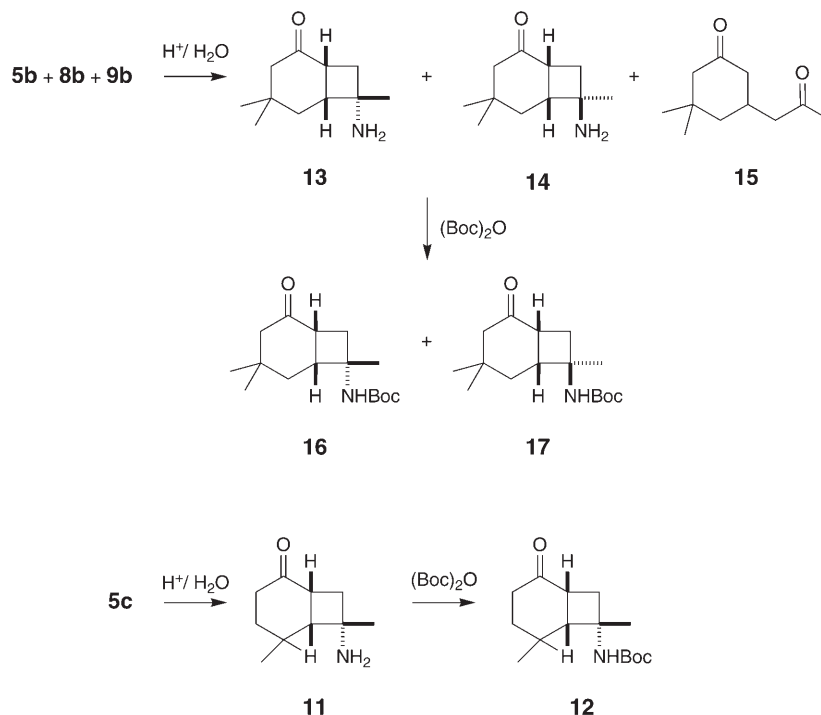
Scheme 3



Chromatographic workup of the product mixtures turns out to be cumbersome, because the isocyanates decompose on prolonged contact with SiO_2 , and, therefore, the crude photoproducts themselves were hydrolyzed to the corresponding amines. On refluxing for 5 h in a mixture of concentrated HCl and benzene [7], **5c** affords the – now *cis*-fused – amine **11**, isolated by separating the organic phase, adding base to the aqueous phase, and extracting with Et_2O , and after that converted into the *N*-Boc derivative **12**. On similar treatment, the mixture of **5b**, **8b**, and **9b** affords a 2.5:1 mixture of amines **13** and **14**, but, in addition, evaporation of the – original – benzene phase affords a product **15** (10% overall yield), identified as 3,3-dimethyl-5-(2-oxopropyl)cyclohexanone. The amines were converted into *N*-Boc derivatives **16** and **17**, and these were then separated by chromatography (Scheme 4). Expectedly, the $^1\text{H-NMR}$ spectra of the *N*-Boc derivatives exhibit strong signal broadening due to the restricted rotation of the carbamate group. Under the same conditions, the isocyanates derived from pyranone **3a** afford large amounts of products resulting from tetrahydropyran ring opening.

3. Discussion. – The overall outcome of photocycloadditions of **3** to **4** resembles that of analogous irradiations of cyclohexenones in the presence of 2-methylpropene [8][9]. Indeed, for the parent cyclohexenone itself, the *trans*-fused *HT*-cycloadduct is the major and the *cis*-fused *HH*-cycloadduct a minor product, whereas, for **3a**, *HT*-products are formed selectively, the ratio of *cis*- to *trans*-fused cycloadduct being roughly 3:2 [10]. Interestingly, no products stemming from H-atom transfer in a intermediate biradical, *i.e.*, so-called ene-type products [11], are observed in the reactions of **3** and **4**. The NMR data of cycloadducts **5a**–**10a** correspond fully to those observed for the photodimers of **3a** [12]. The fact that the *trans*-ring fusion in the bicyclic isocyanates is transformed into a *cis*-fusion in the corresponding amines does not come as a surprise, as epimerization of a (thermodynamically less stable) *trans*-fused bicyclo[4.2.0]octan-2-one to the less strained *cis*-fused diastereoisomer is easily achieved both by base and by acid catalysis. Finally, the conversion of **9b** into **15** can be readily explained by a sequence involving 1) hydrolysis of the isocyanate to an amine,

Scheme 4



2) ring opening of the cyclobutane *via retro-Mannich* reaction [13], and 3) hydrolysis of the iminium ion to a C=O group.

In summary, we have shown, for the first time, that alkenyl isocyanates can be used as (photoinert) reaction component in light-induced enone–alkene cycloadditions, and that the resulting cycloadducts can be converted into the corresponding amino-bicyclooctanones.

Experimental Part

1. *General*. Photolyses were run in a *Rayonet RPR-100* photoreactor equipped with (16) 350 nm lamps and solvents of spectrophotometric grade. Column chromatography (CC): silica gel 60 (*Merck*; 230–400 mesh). 1H - and ^{13}C -NMR spectra (including two-dimensional plots): in $CDCl_3$ at 500.13 and 125.8 MHz, resp., δ in ppm, J in Hz. GC/EI-MS: at 70 eV; 30 m *SE-30* capillary column.

2. *Starting Materials*. 2,3-Dihydro-2,2-dimethyl-4H-pyran-4-one (**3a**) [14], 5,5-dimethylcyclohex-2-enone (**3b**) [15], 4,4-dimethylcyclohex-2-enone (**3c**) [16], and propen-2-yl isocyanate (**4**) [17] were synthesized according to the references indicated.

3. *Photolyses*. Ar-Degassed solns. of **3** (2 mmol) and **4** (20 mmol) in benzene (5 ml) were irradiated for 6–8 h. Total conversion to cycloadducts for **3a** and **3b**, but only 50% conversion of **3c** was achieved, as monitored by both GC and 1H -NMR. After evaporation of the solvent, the composition of the product mixture was determined by NMR spectroscopy.

3.1. *Irradiation of 3a*. The product mixture (>95% overall yield) consisted of the *trans*-fused *HT*-cycloadducts **5a** (18%) and **6a** (8%), of the *cis*-fused *HT*-cycloadducts **7a** (23%) and **8a** (30%), and of the *cis*-fused *HH*-cycloadducts **9a** (17%) and **10a** (4%).

Data of the Major Product (1 α ,6 α ,8 α)-8-Isocyanato-3,3,8-trimethyl-2-oxabicyclo[4.2.0]octan-5-one (8a): $^1\text{H-NMR}$: 4.43 (*dd*, $J = 6.5, 2.0$, H–C(1)); 3.08 (*ddd*, $J = 10.0, 7.5, 6.5$, H–C(6)); 2.51, 2.31 (*AB*, $J = 16.7$, $\text{CH}_2(4)$); 2.44–2.33 (*m*, $\text{CH}_2(7)$); 1.37, 1.36, 1.28 (3*s*, 3 Me). $^{13}\text{C-NMR}$: 208.6 (*s*, C(5)); 122.5 (*s*, N=C=O); 77.3 (*d*, C(1)); 76.0 (*s*, C(3)); 59.8 (*s*, C(8)); 50.2 (*t*, C(4)); 38.6 (*d*, C(6)); 36.8 (*t*, C(7)); 30.3, 30.0, 25.3 (3*q*, Me). *CI-MS*: 210 (48, $[\text{M} + \text{H}]^+$), 111 (100).

3.2. *Irradiation of 3b.* The product mixture (> 90% overall yield) consisted of the *trans*-fused *HT*-cycloadduct **5b** (57%), the *cis*-fused *HT*-cycloadduct **8b** (18%), and the *cis*-fused *HH*-cycloadduct **9b** (25%).

Data of the Major Product (1 α ,6 β ,7 α)-7-Isocyanato-4,4,7-trimethylbicyclo[4.2.0]octan-2-one (5b): $^1\text{H-NMR}$: 3.07 (*ddd*, $J = 13.2, 10.8, 6.0$, H–C(1)); 2.43 (*d*, $J = 13.2$, $\text{H}_{\text{ax}}\text{--C}(3)$); 2.30 (*ddd*, $J = 13.2, 11.5, 3.5$, H–C(6)); 2.14 (*dd*, $J = 10.8, 10.7$, $\text{H}_{\text{ax}}\text{--C}(8)$); 1.95 (*dd*, $J = 10.7, 6.0$, $\text{H}_{\text{eq}}\text{--C}(8)$); 1.83 (*dd*, $J = 13.2, 1.0$, $\text{H}_{\text{eq}}\text{--C}(3)$); 1.73 (*dd*, $J = 12.3, 11.5$, $\text{H}_{\text{ax}}\text{--C}(5)$); 1.57 (*ddd*, $J = 12.3, 3.5, 1.0$, $\text{H}_{\text{eq}}\text{--C}(5)$); 1.51, 1.14, 1.00 (3*s*, 3 Me). $^{13}\text{C-NMR}$: 204.2 (*s*, C(2)); 123.4 (*s*, N=C=O); 63.1 (*s*, C(7)); 54.8 (*t*, C(3)); 53.7 (*d*, C(6)); 47.7 (*d*, C(1)); 38.6 (*s*, C(4)); 38.4 (*t*, C(5)); 37.2 (*t*, C(8)); 31.9, 28.0, 27.5 (3*q*, 3 Me). *EI-MS*: 207 (3, M^+), 68 (100).

3.3. *Irradiation of 3c.* The product mixture consisted mainly of **5c** (40%) and of unreacted **3c** (40–50%).

Data of (1 α ,6 β ,7 α)-7-Isocyanato-5,5,7-trimethylbicyclo[4.2.0]octan-2-one (5c): $^1\text{H-NMR}$: 3.33 (*ddd*, $J = 13.9, 9.7, 6.3$, H–C(1)); 2.43–2.37 (*m*, $\text{CH}_2(3)$); 2.20 (*dd*, $J = 10.8, 9.7$, $\text{H}_{\text{ax}}\text{--C}(8)$); 2.03 (*dd*, $J = 10.8, 6.3$, $\text{H}_{\text{eq}}\text{--C}(8)$); 1.87 (*d*, $J = 13.9$, H–C(6)); 1.76–1.70 (*m*, $\text{CH}_2(4)$); 1.51, 1.33, 1.05 (3*s*, 3 Me). $^{13}\text{C-NMR}$: 209.2 (*s*, C(2)); 123.4 (*s*, N=C=O); 64.1 (*s*, C(7)); 62.7 (*d*, C(6)); 45.5 (*t*, C(4)); 41.9 (*d*, C(1)); 38.9 (*t*, C(3)); 38.5 (*t*, C(8)); 34.0 (*s*, C(5)); 29.0, 27.4, 27.3 (3*q*, 3 Me). *EI-MS*: 207 (4, M^+), 125 (100).

4. *Hydrolysis of Isocyanates.* The crude isocyanates were refluxed in a mixture of benzene (1 ml) and conc. aq. HCl (1 ml) for 5 h. After separation of the aq. phase, the org. phase was extracted three times with 3*N* HCl (1 ml). NaOH was added to the combined aq. phases until pH *ca.* 10, which were then extracted with pentane (6 \times 2 ml). The org. phase was dried (KOH), and the solvent was evaporated to afford the crude amine(s), which were further treated with (Boc) $_2$ O to afford the corresponding *N*-Boc derivatives [18].

4.1. *Hydrolysis of 5c. (1 α ,6 α ,7 β)-7-Amino-5,5,7-trimethylbicyclo[4.2.0]octan-2-one (11):* 87 mg, 50% was obtained. $^1\text{H-NMR}$: 2.60 (*ddd*, $J = 8.8, 6.7, 3.8$, H–C(1)); 2.50 (*dt*, $J = 12.9, 5.3$, $\text{H}_{\text{ax}}\text{--C}(4)$); 2.32 (*dt*, $J = 14.5, 6.0$, $\text{H}_{\text{ax}}\text{--C}(3)$); 2.28 (*m*, $\text{H}_{\text{eq}}\text{--C}(3)$); 2.13 (*dd*, $J = 8.8, 2.0$, H–C(6)); 1.89–1.88 (*m*, $\text{CH}_2(8)$); 1.44 (*dddd*, $J = 12.9, 8.2, 4.4, 2.0$, $\text{H}_{\text{eq}}\text{--C}(4)$); 1.20, 0.99, 0.97 (3*s*, 3 Me). $^{13}\text{C-NMR}$: 214.0 (*s*, C(2)); 58.2 (*s*, C(7)); 55.1 (*d*, C(6)); 38.8 (*d*, C(1)); 37.1 (*t*, C(8)); 36.0 (*t*, C(3)); 35.2 (*t*, C(4)); 30.1 (*s*, C(5)); 31.9, 27.8, 24.8 (3*q*, 3 Me). *EI-MS*: 181 (0.4, M^+), 57 (100). The corresponding (*tert*-Butoxy)carbonyl (Boc) derivative **12** was recrystallized from AcOEt/hexane; m.p. 68–70°.

4.2. *Hydrolysis of the Mixture 5b, 8b, and 9b.* A 2.5 : 1 mixture of the diastereoisomeric amines **16** and **17** was obtained, which were directly converted into their *N*-Boc derivatives. CC (SiO $_2$; Et $_2$ O)/pentane 3 : 1 afforded as a first fraction 50 mg (9%) of the minor product (R_f 0.39) (*1 α ,6 α ,7 α)-7-[(*tert*-butoxy)carbonyl]amino]-4,4,7-trimethylbicyclo[4.2.0]octan-2-one (17) as a light yellow oil. The second fraction contained 129 mg (23%) of the major product (R_f 0.34) (*1 α ,6 α ,7 β)-7-[(*tert*-butoxy)carbonyl]amino]-4,4,7-trimethylbicyclo[4.2.0]octan-2-one (16).**

Data of 17: $^1\text{H-NMR}$: 3.03 (*ddd*, $J = 9.5, 9.0, 8.5$, H–C(1)); 2.93 (*br. m*, H–C(6)); 2.38 (*br. m*, 1 H of $\text{CH}_2(8)$); 2.20 (*m*, $\text{CH}_2(3)$); 2.18 (*br. m*, 1 H of $\text{CH}_2(8)$); 1.71 (*br. m*, 1 H of $\text{CH}_2(5)$); 1.52 (*br. m*, 1 H of $\text{CH}_2(5)$); 1.44 (*s*, Me_3C); 1.26, 1.06, 0.95 (3*s*, 3 Me). $^{13}\text{C-NMR}$: 213.4 (*s*, C(2)); 156.9 (*s*, CO of Boc); 79.3 (*s*, Me_3CO); 55.2 (*s*, C(7)); 52.8 (*t*, C(3)); 42.9 (*d*, C(6)); 38.4 (*d*, C(1)); 36.4 (*t*, C(8)); 35.8 (*t*, C(5)); 33.7 (*s*, C(4)); 32.3 (*q*, Me); 28.9 (*q*, Me_3C); 26.8 (*q*, 2 Me).

Data of 16: M.p. 110–112°. $^1\text{H-NMR}$: 2.80 (*ddd*, $J = 9.5, 9.0, 8.5$, H–C(1)); 2.63 (*br. m*, H–C(6)); 2.27, 2.10 (*AB*, $J = 16.1$, $\text{CH}_2(3)$); 2.26, 2.16 (2 *br. m*, $\text{CH}_2(8)$); 1.71 (*br. m*, 1 H of $\text{CH}_2(5)$); 1.69 (*br. m*, 1 H of $\text{CH}_2(5)$); 1.55 (*s*, Me); 1.44 (*s*, Me_3C); 1.02, 0.85 (2*s*, 2 Me). $^{13}\text{C-NMR}$: 213.1 (*s*, C(2)); 156.9 (*s*, CO of Boc); 79.2 (*s*, Me_3CO); 52.1 (*t*, C(3)); 51.5 (*s*, C(7)); 43.4 (*d*, C(6)); 37.5 (*d*, C(1)); 36.8 (*t*, C(8)); 34.8 (*s*, C(4)); 33.9 (*t*, C(5)); 31.4 (*q*, Me); 28.9 (*q*, Me_3C); 25.2 (*q*, 2 Me).

After drying the original benzene phase (MgSO $_4$) and subsequent evaporation of the solvent, 36 mg (10%) of 3,3-dimethyl-5-(2-oxopropyl)cyclohexanone (**15**) as colorless liquid were obtained. $^1\text{H-NMR}$:

2.46–2.43 (*m*, H–C(5), CH₂(1')); 2.37 (*dd*, *J* = 13.3, 6.3, H_{eq}–C(6)); 2.16 (*d*, *J* = 13.3, H_{ax}–C(2)); 2.14 (*s*, Me(3')); 2.10 (*dd*, *J* = 13.3, 2.2, H_{eq}–C(2)); 1.92 (*dd*, *J* = 13.3, 12.6, H_{ax}–C(6)); 1.63 (*ddd*, *J* = 13.6, 5.9, 2.2, H_{eq}–C(4)); 1.31 (*dd*, *J* = 13.6, 12.6, H_{ax}–C(4)); 1.06, 0.92 (2*s*, 2 Me). ¹³C-NMR: 210.7 (*s*, C(1)); 207.2 (*s*, C(2')); 54.3 (*t*, C(2)); 50.4 (*t*, C(1')); 46.6 (*t*, C(6)); 44.7 (*t*, C(4)); 35.0 (*s*, C(3)); 31.9 (*q*, Me); 30.4 (*q*, Me(3')); 30.2 (*d*, C(5)); 25.6 (*q*, Me). EI-MS: 182 (12, *M*⁺), 43 (100).

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